

## A New and Facile Synthesis of 1*H*-Indazoles

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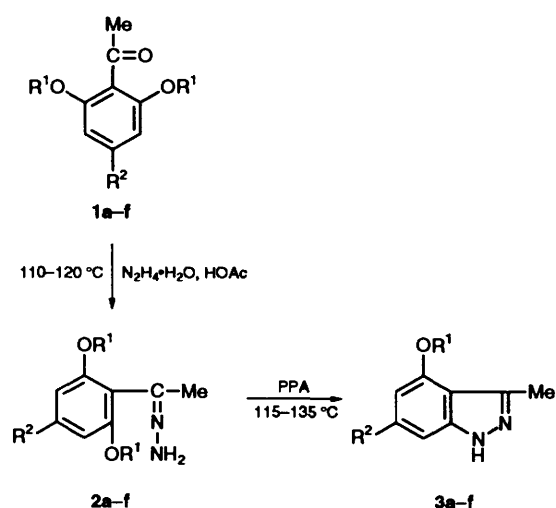
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A new and convenient procedure is described for the synthesis of 1*H*-indazoles by cyclization of 2,6-dialkoxyacetophenone hydrazones in the presence of polyphosphoric acid (PPA).

Of the three classical methods for the preparation of 1*H*-indazoles the most frequently used is the diazotization of suitably substituted anilines bearing a hydrocarbon group at the *ortho*-position.<sup>1</sup> A similar reaction occurs when *N*-nitroso-2-methylanilines are heated in the presence of sodium carbonate,<sup>2</sup> and treatment of *o*-chloro-aromatic ketones with a nitro substituent *para* to the chloro substituent with arylhydrazines also gives 1-aryl-1*H*-indazoles.<sup>3</sup> These methods have several limitations as regards the reaction conditions; therefore, several improved syntheses<sup>4–10</sup> have been proposed or developed. However, as far as we know, an efficient synthesis of 1*H*-indazoles in reasonable yield has not yet been reported. We describe here a facile one-pot synthesis of 1*H*-indazoles in high yield.



1–3	R <sup>1</sup>	R <sup>2</sup>
a	H	H
b	Me	H
c	Et	H
d	Pr	H
e	Bu	H
f	Me	OMe

Treatment of 2,6-dialkoxy (or hydroxy) acetophenones **1a–f** with hydrazine hydrate in the presence of HOAc at 110–120 °C results in hydrazones **2a–f**. After cooling, without isolation of **2a–f**, the addition of PPA followed by stirring at 110–135 °C for about 20 min, gives 3-methyl-4-alkoxy(or hydroxy)-1*H*-indazoles **3a–f**† as the final products.

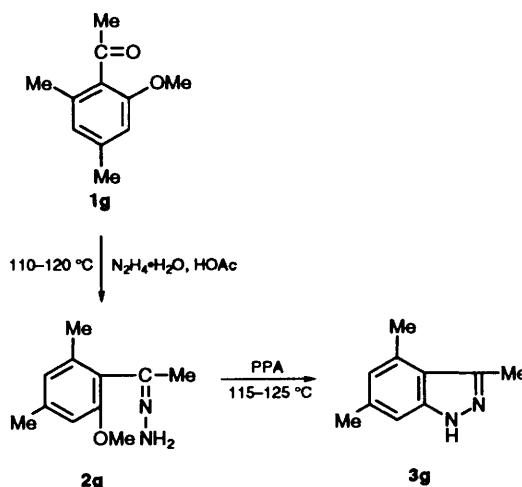
Application of this method to 2-methoxy-4,6-dimethylacetophenone **1g**, gives 3,4,6-trimethyl-1*H*-indazole **3g**.

In summary, this new procedure offers several advantages: the

**Table 1** Synthetic conditions, yields and m.p.s for compounds **3a–g**

Compound	T/°C	t/min	Yield (%) <sup>a</sup>	M.p. (°C) <sup>b</sup>
<b>3a</b>	115–125	25	34	206–208
<b>3b</b>	120–130	20	87	128–130
<b>3c</b>	120–135	25	80	130–131
<b>3d</b>	120–135	30	62	100–102
<b>3e</b>	120–130	30	60	64–65
<b>3f</b>	120–135	25	65	166–168
<b>3g</b>	115–125	25	73	206–208

<sup>a</sup> Yield of isolated product. <sup>b</sup> Uncorrected.



easy availability of the reagents, reasonable yield, short reaction time and convenience.

### Experimental

**4-Acetoxy-3-methyl-1*H*-indazole **3a**: Typical Procedure A.**—2,6-Dihydroxyacetophenone **1a** (1.52 g, 10 mmol), 85% hydrazine hydrate (1.20 g, 20 mmol) and HOAc (10 drops) were placed in a flask (50 cm<sup>3</sup> volume). The mixture was stirred at 110–120 °C (oil bath heating) and the reaction was monitored by TLC for 15 min. After cooling, PPA (12 g) was added to the reaction mixture, which was then stirred at 115–125 °C for 20 min during which the intermediate **2a** disappeared. After cooling, acetic anhydride (1.2 g, 15 mmol) was added to the reaction mixture, which was stirred at 120 °C for 25 min. After cooling, ice-water was added to the reaction mixture. The product was extracted with EtOAc (3 × 40 cm<sup>3</sup>) and the combined extracts were washed with water (2 × 20 cm<sup>3</sup>), dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated to give a crystalline residue. Chromatography on silica gel using light petroleum (b.p. 60–90 °C) and EtOAc (4:1) as eluent yielded **3a** (0.65 g, 34%); m.p. 206–208 °C.

† As 3-methyl-1*H*-indazol-4-ol is very soluble in water it could not easily be extracted from water and was therefore acetylated with acetic anhydride to give 4-acetoxy-3-methyl-1*H*-indazole.

**Table 2** Elemental analyses and spectroscopic characterizations for **3a-g**

Compound	Molecular formula	Found <sup>a</sup> (Required) (%)			$\nu_{\max}/\text{cm}^{-1}$ <sup>b</sup>	$m/z$ <sup>c</sup>	$\delta_{\text{H}}[\text{CDCl}_3; (\text{CH}_3)_4\text{Si}]$ <sup>d</sup>
		C	H	N			
<b>3a</b>	C <sub>10</sub> H <sub>10</sub> N <sub>2</sub> O <sub>2</sub> (190.20)	63.25 (63.14)	5.12 5.30	14.38 14.73	3165, 1760, 1627 1597, 1535, 1204, 1153, 820	190 (M <sup>+</sup> ), 175 (b), 147, 119	2.64 (3 H, s, OCCH <sub>3</sub> ), 2.75 (3 H, s, CH <sub>3</sub> ), 5.85 (1 H, br, NH), 6.92–7.70 (3 H, m, ArH)
<b>3b</b>	C <sub>9</sub> H <sub>10</sub> N <sub>2</sub> O (162.19)	66.64 (66.67)	6.47 6.21	17.16 17.28	3171, 1623, 1596, 1525, 1354, 1257, 1114, 995, 847	162 (M <sup>+</sup> , b), 147, 131, 119	2.80 (3 H, s, CH <sub>3</sub> ), 4.07 (3 H, s, OCH <sub>3</sub> ), 6.60–7.67 (3 H, m, ArH), 8.56 (1 H, br, NH)
<b>3c</b>	C <sub>10</sub> H <sub>12</sub> N <sub>2</sub> O (176.22)	68.06 (68.15)	7.07 6.81	15.97 15.90	3166, 1622, 1597, 1525, 1373, 1261, 1122, 774	176 (M <sup>+</sup> ), 148 (b), 131, 119	1.64 (3 H, t, OCH <sub>2</sub> CH <sub>3</sub> ), 2.95 (3 H, s, CH <sub>3</sub> ), 4.52 (2 H, q, OCH <sub>2</sub> CH <sub>3</sub> ), 6.94– 8.00 (3 H, m, ArH), 9.60 (1 H, br, NH)
<b>3d</b>	C <sub>11</sub> H <sub>14</sub> N <sub>2</sub> O (190.24)	69.61 (69.64)	7.00 7.42	14.60 14.73	3166, 1623, 1596, 1524, 1373, 1261, 1100, 768	190 (M <sup>+</sup> ), 148 (b), 147, 119	1.07 (3 H, t, OCH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> ), 1.84 (2 H, m, OCH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> ), 2.62 (3 H, s, CH <sub>3</sub> ), 3.87 (2 H, t, OCH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> ), 6.02–7.04 (3 H, m, ArH), 9.15 (1 H, br, NH)
<b>3e</b>	C <sub>12</sub> H <sub>16</sub> N <sub>2</sub> O (204.27)	70.29 (70.55)	7.66 7.90	13.63 13.72	3167, 1622, 1597, 1524, 1373, 1263, 1110, 772	204 (M <sup>+</sup> ), 176, 148 (b), 147, 119	0.80–2.15 (7 H, m, OCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> ), 2.65 (3 H, s, CH <sub>3</sub> ), 3.90 (2 H, t, OCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> ), 6.06–7.05 (3 H, m, ArH), 9.50 (1 H, br, NH)
<b>3f</b>	C <sub>10</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub> (192.22)	62.52 (62.28)	6.08 6.29	14.36 14.57	3182, 1636, 1602, 1531, 1392, 1202, 1144, 804	192 (M <sup>+</sup> , b) 117, 149, 134, 119	2.87 (3 H, s, CH <sub>3</sub> ), 4.15 (3 H, m, s, OCH <sub>3</sub> ) 4.24 (3 H, s, OCH <sub>3</sub> ), 6.67–6.94 (2 H, m, ArH), 7.92 (1 H, s, NH)
<b>3g</b>	C <sub>10</sub> H <sub>12</sub> N <sub>2</sub> (160.22)	74.94 (74.96)	7.76 7.55	17.85 17.49	3176, 1622, 1597, 1445, 1344, 1231, 990, 833	160 (M <sup>+</sup> , b) 159, 145, 115	2.64 (3 H, s, CH <sub>3</sub> ), 2.88 (3 H, s, Ar-CH <sub>3</sub> ), 2.94 (3 H, s, Ar-CH <sub>3</sub> ), 7.45 (2 H, d, ArH), 7.87 (1 H, s, NH)

<sup>a</sup> Analysed on a Perkin-Elmer 240C element analytical meter. <sup>b</sup> Recorded on a Nicolet-170 SX-FT-IR spectrophotometer (KBr). <sup>c</sup> Recorded on a VG-ZAB-HS mass spectrometer (EI). <sup>d</sup> Recorded on a JEOL PMX-60SI spectrometer (60 MHz).

**4-Methoxy-3-methyl-1H-indazole 3b: Typical Procedure B.**—A mixture of 2,6-dimethoxyacetophenone **1b** (0.90 g, 5 mmol), 85% hydrazine hydrate (0.60 g, 10 mmol) and HOAc (10 drops) was stirred at 110–120 °C, and monitored by TLC. After 25 min the intermediate **1b** had disappeared. After cooling, PPA (10 g) was added to the mixture, which was then stirred at 120–135 °C for 20 min, during which time the intermediate **2b** disappeared. After cooling, ice-water was added to the mixture. The product was extracted with EtOAc (3 × 40 cm<sup>3</sup>), and the combined extracts were washed with water, dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated to give a brown crystalline crude product which was purified as above to give white needles of product **3b**; yield (0.71 g, 87%); m.p. 128–130 °C. Compounds **3c-g** were obtained by a similar procedure as **B** from **1c-g**.

Compounds **3a-g** were fully characterized by IR, <sup>1</sup>H NMR, MS and elemental analysis, the data being listed in Tables 1 and 2.

## References

- 1 R. Huisgen and H. Nakaten, *Liebigs Ann. Chem.*, 1954, **586**, 84.
- 2 P. Jacobson and L. Huber, *Ber. Dtsch. Chem. Ges.*, 1908, **41**, 660.
- 3 W. Borsche and W. Scriba, *Liebigs Ann. Chem.*, 1939, **540**, 83.
- 4 N. Virona, G. Cusmano, G. Maculoso, V. Frenna and M. Ruccia, *J. Heterocycl. Chem.*, 1979, **16**, 783.
- 5 W. A. F. Gladstone and R. O. C. Norman, *J. Chem. Soc.*, 1965, 3048.
- 6 T. Yamazaki, G. Baum and H. Sheter, *Tetrahedron Lett.*, 1974, 4421.
- 7 E. G. Abbad, M. T. G. Lopez, G. G. Munoz and M. Stud, *J. Heterocycl. Chem.*, 1976, **13**, 1241.
- 8 K. H. Mayer, D. Lauerer and H. Heitzer, *Synthesis*, 1977, 804.
- 9 M. P. Kausik, *J. Org. Chem.*, 1982, **47**, 3503.
- 10 S. Matsugo, M. Saito and A. Takamizawa, *Synthesis*, 1983, 482.

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